



Year: 2019

**Crystal structures and Hirshfeld surface analysis of
2-(adamantan-1-yl)-5-(4-fluorophenyl)-1,3,4-oxadiazole and
2-(adamantan-1-yl)-5-(4-chlorophenyl)-1,3,4-oxadiazole**

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Abstract: The crystal structures of the title adamantane-oxadiazole hybrid compounds, C₁₈H₁₉FN₂O (I) and C₁₈H₁₉ClN₂O (II), are built up from an adamantane unit and a halogenophenyl ring, [X = F (I), Cl (II) in position 5 on the central 1,3,4-oxadiazole unit. The molecular structures are very similar, only the relative orientation of the halogenophenyl ring in comparison with the central five membered ring differs slightly. In the crystals of both compounds, molecules are linked by pairs of C-H center dot center dot N hydrogen bonds, forming inversion dimers with R-2(2)(12) ring motifs. In (I) the dimers are connected by C-H center dot center dot center dot F interactions, forming slabs lying parallel to the bc plane. In (II), the dimers are linked by C-H center dot center dot center dot pi and offset pi-pi interactions [interplanar distance = 3.4039 (9) angstrom], forming layers parallel to (10 1) over bar).

DOI: <https://doi.org/10.1107/s2056989019004651>

Posted at the Zurich Open Repository and Archive, University of Zurich

ZORA URL: <https://doi.org/10.5167/uzh-182129>

Journal Article

Published Version



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Originally published at:

Al-Wahaibi, Lamya H; Alsfolk, Aisha; El-Emam, Ali A; Blacque, Olivier (2019). Crystal structures and Hirshfeld surface analysis of 2-(adamantan-1-yl)-5-(4-fluorophenyl)-1,3,4-oxadiazole and 2-(adamantan-1-yl)-5-(4-chlorophenyl)-1,3,4-oxadiazole. *Acta Crystallographica. Section E: Structure Reports Online*, 75(5):611-615.

DOI: <https://doi.org/10.1107/s2056989019004651>

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Received 22 March 2019

Accepted 5 April 2019

Edited by H. Stoeckli-Evans, University of Neuchâtel, Switzerland

Keywords: crystal structure; adamantyl derivatives; 1,3,4-oxadiazoles; C—H···N hydrogen bonds; C—H···F interactions; Hirshfeld surface analysis.

CCDC references: 1908204; 1908203

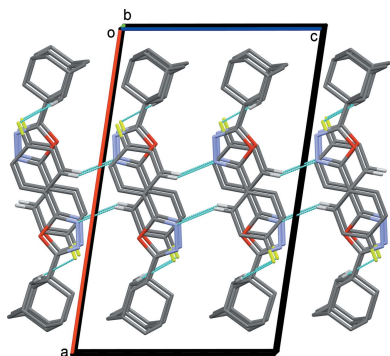
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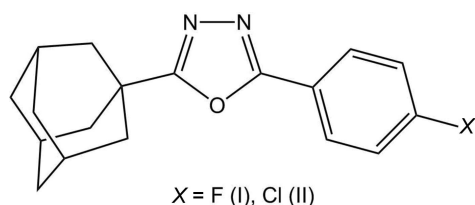
The crystal structures of the title adamantane-oxadiazole hybrid compounds, C₁₈H₁₉FN₂O (I) and C₁₈H₁₉ClN₂O (II), are built up from an adamantane unit and a halogenophenyl ring, [X = F (I), Cl (II)], in position 5 on the central 1,3,4-oxadiazole unit. The molecular structures are very similar, only the relative orientation of the halogenophenyl ring in comparison with the central five-membered ring differs slightly. In the crystals of both compounds, molecules are linked by pairs of C—H···N hydrogen bonds, forming inversion dimers with R₂²(12) ring motifs. In (I) the dimers are connected by C—H···F interactions, forming slabs lying parallel to the *bc* plane. In (II), the dimers are linked by C—H··· π and offset π – π interactions [interplanar distance = 3.4039 (9) Å], forming layers parallel to (10 $\bar{1}$).

1. Chemical context

Considerable attention has been devoted to adamantane derivatives, which have long been known for their diverse biological properties (Liu *et al.*, 2011; Lamoureux & Artavia, 2010). In view of the pronounced lipophilicity of the adamantane cage, it has been observed that adamantyl-bearing compounds are characterized by high therapeutic indices (Wanka *et al.*, 2013). Sixty years ago, the first adamantane-based drug, amantadine, was discovered to be an efficient therapy for the treatment of Influenza A infection (Davies *et al.*, 1964; Togo *et al.*, 1968). As a result of intensive research based on adamantane derivatives, the adamantane nucleus was further recognized as the key pharmacophore in several biologically active compounds. Among the major biological activities displayed by adamantane derivatives, the anti-HIV (El-Emam *et al.*, 2004; Burstein *et al.*, 1999; Balzarini *et al.*, 2009), antibacterial (Protopopova *et al.*, 2005; El-Emam *et al.*, 2013; Kadi *et al.*, 2010; Al-Abdullah *et al.*, 2014; Al-Wahaibi *et al.*, 2017), antifungal (Omar *et al.*, 2010), anticancer (Sun *et al.*, 2002), anti-diabetic (Villhauer *et al.*, 2003; Augeri *et al.*, 2005) and antimalarial (Dong *et al.*, 2010) activities are the most interesting. In addition, 1,3,4-oxadiazole derivatives occupy a unique place in the field of medicinal chemistry as pharmacophores or auxophores possessing diverse pharmacological activities including antibacterial (Prakash *et al.*, 2010; Ogata *et al.*, 1971; Kadi *et al.*, 2007), anticancer (Zhang *et al.*,



2014), antiviral (Wu *et al.*, 2015) and anti-inflammatory (Bansal *et al.*, 2014) activities. We report herein on the crystal structure determinations of the title adamantane-oxadiazole hybrid molecules 2-(adamantan-1-yl)-5-(4-fluorophenyl)-1,3,4-oxadiazole (I) and 2-(adamantan-1-yl)-5-(4-chlorophenyl)-1,3,4-oxadiazole (II). The crystal structure of the 4-bromophenyl derivative has been reported previously (Alzoman *et al.*, 2014), and after examination of the deposited CIF and transformation of the space group, from $P2_1/c$ to $P2_1/n$, it is found to be isotopic with compound (II).



2. Structural commentary

Compounds (I) and (II), are built up from a central 1,3,4-oxadiazole unit, an adamantane unit and a halogenophenyl group (Figs. 1 and 2, respectively). The C–N bonds in the oxadiazole rings have double-bond character [C7=N1 = 1.279 (5) and 1.292 (3) Å, and C8=N2 = 1.288 (5) and 1.288 (3) Å in (I) and (II), respectively], while the N–N and C–O bonds exhibit single-bond character [N1–N2 = 1.408 (4) and 1.417 (3) Å, C7–O1 = 1.366 (4) and 1.360 (2) Å, and C8–O1 = 1.369 (4) and 1.359 (2) Å in (I) and (II), respectively]. These geometrical parameters are very similar

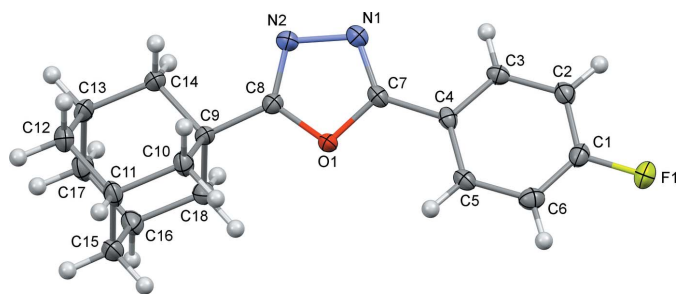


Figure 1
Molecular structure of compound (I), with the atom labelling and displacement ellipsoids drawn at the 50% probability level.

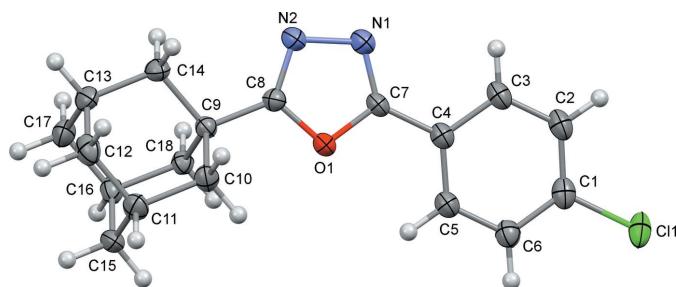


Figure 2
Molecular structure of compound (II), with the atom labelling and displacement ellipsoids drawn at the 50% probability level.

Table 1
Hydrogen-bond geometry (Å, °) for (I).

Cg1 is the centroid of the C1–C6 ring.

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
C3–H3...N1 ⁱ	0.95	2.56	3.383 (5)	146
C18–H18A...F1 ⁱⁱ	0.99	2.47	3.415 (5)	159

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$.

Table 2
Hydrogen-bond geometry (Å, °) for (II).

Cg1 is the centroid of the C1–C6 ring.

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
C3–H3...N1 ⁱ	0.95	2.61	3.386 (3)	139
C12–H12A...Cg1 ⁱⁱ	0.99	2.73	3.680 (3)	160

Symmetry codes: (i) $-x + 1, -y, -z + 1$; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

to those observed for similar compounds; see §5. *Database survey*.

As seen in Fig. 3, the molecular structures of compounds (I) and (II) are very similar. The largest difference is highlighted by the structural overlay plot, and comes from the relative orientation of the halogenophenyl group with respect to the oxadiazole ring. In compound (II), the rings are almost coplanar with their mean planes being inclined to each other by 9.5 (1)°, while in compound (I) the equivalent dihedral angle is 20.8 (2)°.

3. Supramolecular features

In the crystals of both compounds, molecules are linked by pairs of C–H...N hydrogen bonds, forming inversion dimers with $R_2^2(12)$ ring motifs (Tables 1 and 2, respectively). In the crystal of (I), the dimers are connected by C–H...F interactions, forming slabs lying parallel to the *bc* plane (Fig. 4 and Table 1). In the crystal of (II), the dimers are linked by C–H... π and offset π – π interactions, forming layers lying parallel to the (10 $\bar{1}$) plane; see Fig. 5 and Table 2. The offset π – π interactions involve inversion-related 4-chlorophenyl rings (C1–C6) with an intercentroid distance of 3.687 (1) Å, an interplanar distance of 3.404 (1) Å, and an offset of 1.418 Å. In Fig. 5 these interactions are represented by double-headed red arrows.

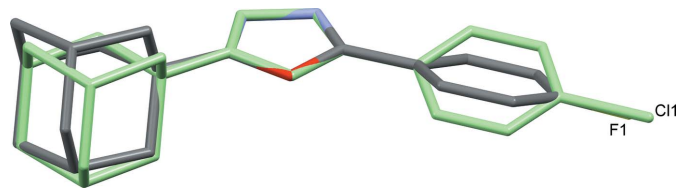


Figure 3
View of the structural overlay of compounds (I) and (II). Compound (I) is drawn according to element type, while compound (II) is drawn in pale green.

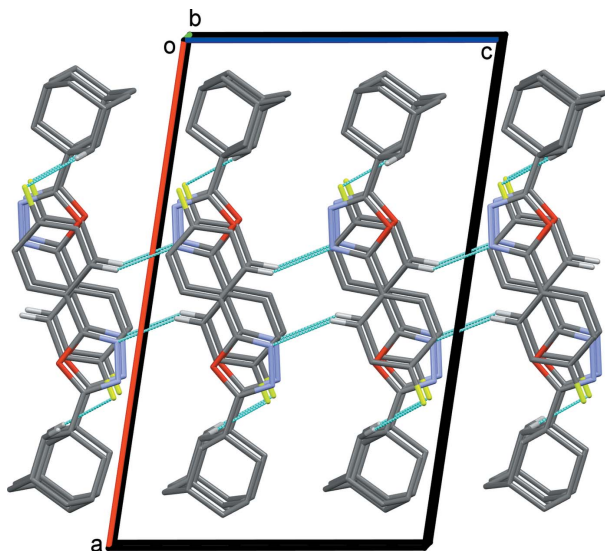


Figure 4
A view along the *b* axis of the crystal packing of compound (I). The hydrogen-bonding interactions (see Table 1) are shown as dashed lines. For clarity, only hydrogen atoms H3 and H18A have been included.

4. Hirshfeld surface analysis

The Hirshfeld surfaces for (I) and (II) mapped over d_{norm} were calculated using *CrystalExplorer 17* (Turner *et al.*, 2017) with the default setting of arbitrary units range. The characteristic bright-red spots near atoms H3, H18A, N1 and F1 (Fig. 6)

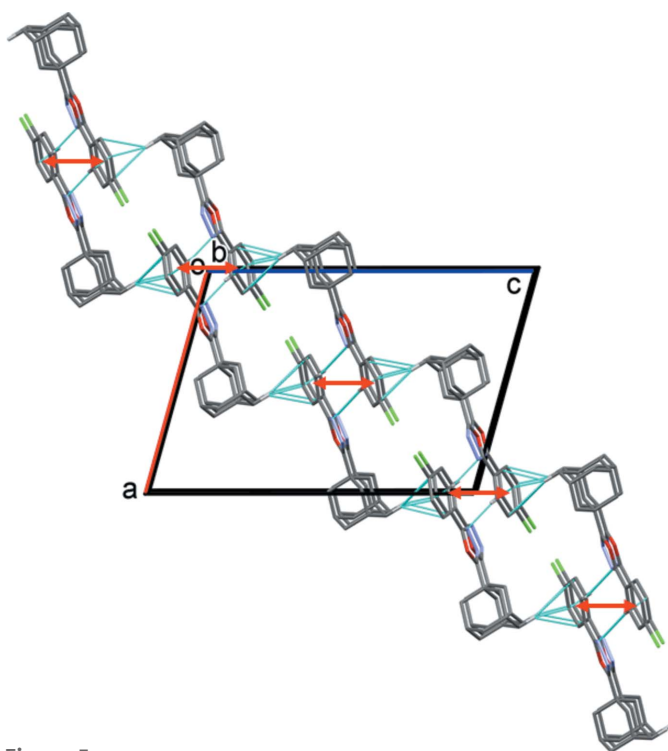


Figure 5
A view along the *b* axis of the crystal packing of compound (II), showing the C–H...N hydrogen bonds and the C–H... π interactions (see Table 2) as dashed lines. The offset π – π interactions are indicated by double-headed red arrows. For clarity, only hydrogen atoms H3 and H12A have been included.

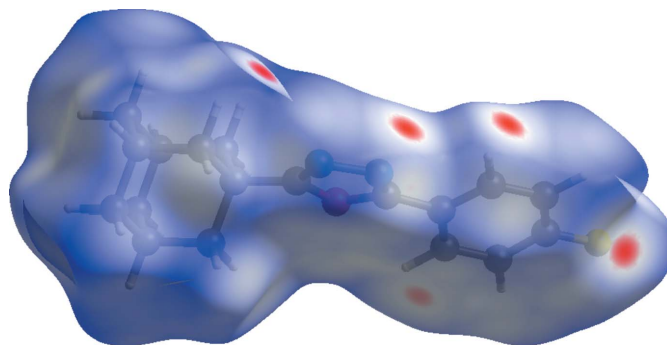


Figure 6
A view of the Hirshfeld surface mapped over d_{norm} for compound (I) over the range -0.138 to 1.364 arbitrary units.

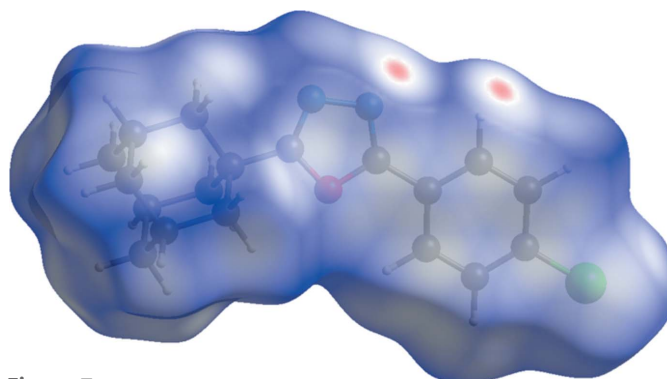


Figure 7
A view of the Hirshfeld surface mapped over d_{norm} for compound (II) over the range -0.203 to 1.273 arbitrary units.

confirm the previously mentioned C3–H3...N1ⁱ and C18–H18A...F1ⁱⁱ [symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$] interatomic contacts in the crystal packing of (I). As expected, the same bright-red spots are observed near atoms H3 and N1 on the Hirshfeld surface of (II); see Fig. 7. The Hirshfeld surface mapped over the shape-index property elegantly illustrates the π – π stacking and the C–H... π interactions observed in the crystal packing of (II).

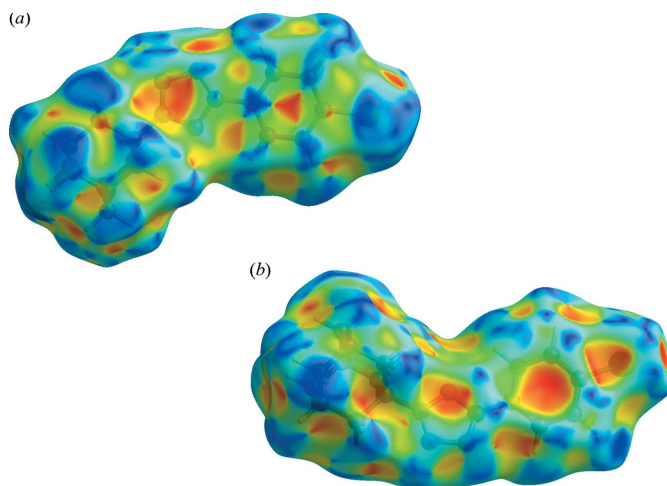


Figure 8
Two views, (a) and (b), of the Hirshfeld surface mapped over the shape-index property for compound (II).

Table 3
Experimental details.

	(I)	(II)
Crystal data		
Chemical formula	C ₁₈ H ₁₉ FN ₂ O	C ₁₈ H ₁₉ ClN ₂ O
<i>M_r</i>	298.35	314.80
Crystal system, space group	Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>	Monoclinic, <i>P</i> ₂ ₁ / <i>n</i>
Temperature (K)	160	160
<i>a</i> , <i>b</i> , <i>c</i> (Å)	18.2525 (4), 7.07855 (16), 11.2207 (2)	13.08241 (19), 6.49259 (9), 18.5129 (3)
β (°)	98.556 (2)	105.5609 (16)
<i>V</i> (Å ³)	1433.59 (6)	1514.83 (4)
<i>Z</i>	4	4
Radiation type	Cu <i>K</i> α	Cu <i>K</i> α
μ (mm ⁻¹)	0.78	2.25
Crystal size (mm)	0.18 × 0.15 × 0.02	0.33 × 0.12 × 0.08
Data collection		
Diffractometer	XtaLAB Synergy, Dualflex, Pilatus 200K	XtaLAB Synergy, Dualflex, Pilatus 200K
Absorption correction	Analytical (<i>CrysAlis PRO</i> ; Rigaku OD, 2019)	Analytical (<i>CrysAlis PRO</i> ; Rigaku OD, 2019)
<i>T</i> _{min} , <i>T</i> _{max}	0.921, 0.990	0.642, 0.870
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	13058, 2903, 2578	14318, 3217, 3052
<i>R</i> _{int}	0.030	0.023
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.625	0.636
Refinement		
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.080, 0.233, 1.23	0.061, 0.167, 1.08
No. of reflections	2903	3217
No. of parameters	199	199
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.47, -0.38	0.81, -0.26

Computer programs: *CrysAlis PRO* (Rigaku OD, 2019), *SHELXT* (Sheldrick, 2015a), *SHELXL* (Sheldrick, 2015b), *Mercury* (Macrae *et al.*, 2008) and *OLEX2* (Dolomanov *et al.*, 2009).

Two views are presented in Fig. 8. The π - π stacking between inversion-related 4-chlorophenyl rings (C1–C6) is indicated by the appearance of small blue regions surrounding a bright-red triangle within the six-membered ring (Fig. 8*a*), while the C12–H12A $\cdots\pi$ (C1–C6)ⁱⁱⁱ interaction [symmetry code: (iii) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$] appears as a large red region within the ring (Fig. 8*b*).

5. Database survey

A search of the Cambridge Structural Database (CSD, version 5.40, February 2019; Groom *et al.*, 2016) for the substructure 2-(adamantan-1-yl)-1,3,4-oxadiazole gave five hits. The crystal structures of three very similar compounds were reported in the last decade, namely 2-(adamantan-1-yl)-5-(4-nitrophenyl)-1,3,4-oxadiazole (CSD refcode LAPVOP; El-Emam *et al.*, 2012), which has an NO₂ group on the phenyl ring (in the *para* position to the oxadiazole ring), 2-(adamantan-1-yl)-5-(4-bromophenyl)-1,3,4-oxadiazole (SOSXIJ; Alzoman *et al.* 2014), which has a Br atom on the phenyl ring (same *para* position) and 2-(adamantan-1-yl)-5-(3-fluorophenyl)-1,3,4-oxadiazole (SIKKAA; Khan *et al.*, 2012), with a 3-fluorophenyl substituent at position 5 on the oxadiazole ring. Two more recently reported structures are 5-(adamantan-1-yl)-1,3,4-oxadiazole-2-thiolato)triphenylphosphinegold(I) (AZECAL; Garcia *et al.*, 2016) and 2-(adamantan-1-yl)-5-[2-(2-methylphenyl)-1,3-thiazol-4-yl]-1,3,4-oxadiazole (XARGEE01; Khan *et al.*, 2016).

The reduced cell of SOSXIJ indicates that it is isotopic with compound (II). Compound LAPVOP resides on a mirror plane, while compound SIKKAA crystallizes with two independent molecules in the asymmetric unit. The geometrical parameters of the oxadiazole rings are similar to those reported above for the title compounds. The 4-substituted phenyl rings are inclined to the oxadiazole ring by 0.0° in LAPVOP (as it lies in a mirror plane), 3.01 and 3.31° in the two independent molecules of SIKKAA and 10.44° in SOSXIJ. In the title compounds the corresponding dihedral angle is 20.8 (2)° for compound (I) and 9.5 (1)° for compound (II).

6. Synthesis and crystallization

Compounds (I) and (II) were synthesized *via* condensation of adamantane-1-carboxylic acid with 4-fluorobenzohydrazide, or 4-chlorobenzohydrazide in the presence of phosphorus oxychloride, as described previously (Kadi *et al.*, 2007). Colourless plate-like crystals of compound (I) and colourless needle-like crystals of compound (II) were obtained by slow evaporation of CHCl₃:EtOH (1:1 *v:v*) solutions at room temperature.

7. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 3. All H atoms were placed in

calculated positions and treated as riding atoms: C—H = 0.95–1.00 Å with $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$.

Funding information

This work was funded by the Deanship of Scientific Research at Princess Nourah bint Abdulrahman University through the Research Group Program (grant No. RGP-1438–0010).

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supporting information

Acta Cryst. (2019). E75, 611-615 [https://doi.org/10.1107/S2056989019004651]

Crystal structures and Hirshfeld surface analysis of 2-(adamantan-1-yl)-5-(4-fluorophenyl)-1,3,4-oxadiazole and 2-(adamantan-1-yl)-5-(4-chlorophenyl)-1,3,4-oxadiazole

Lamya H. Al-Wahaibi, Aisha Alsouk, Ali A. El-Emam and Olivier Blacque

Computing details

For both structures, data collection: *CrysAlis PRO* (Rigaku OD, 2019); cell refinement: *CrysAlis PRO* (Rigaku OD, 2019); data reduction: *CrysAlis PRO* (Rigaku OD, 2019); program(s) used to solve structure: ShelXT (Sheldrick, 2015a); program(s) used to refine structure: *SHELXL* (Sheldrick, 2015b); molecular graphics: *Mercury* (Macrae *et al.*, 2008); software used to prepare material for publication: *OLEX2* (Dolomanov *et al.*, 2009).

2-(Adamantan-1-yl)-5-(4-fluorophenyl)-1,3,4-oxadiazole (I)

Crystal data

C₁₈H₁₉FN₂O

M_r = 298.35

Monoclinic, *P*2₁/*c*

a = 18.2525 (4) Å

b = 7.07855 (16) Å

c = 11.2207 (2) Å

β = 98.556 (2)°

V = 1433.59 (6) Å³

Z = 4

F(000) = 632

D_x = 1.382 Mg m⁻³

Cu *Kα* radiation, λ = 1.54184 Å

Cell parameters from 7588 reflections

θ = 4.9–78.5°

μ = 0.78 mm⁻¹

T = 160 K

Plate, colourless

0.18 × 0.15 × 0.02 mm

Data collection

XtaLAB Synergy, Dualflex, Pilatus 200K diffractometer

Radiation source: micro-focus sealed X-ray tube, PhotonJet (Cu) X-ray Source

Mirror monochromator

Detector resolution: 5.8140 pixels mm⁻¹

ω scans

Absorption correction: analytical (CrysAlisPro; Rigaku OD, 2019)

T_{min} = 0.921, *T_{max}* = 0.990

13058 measured reflections

2903 independent reflections

2578 reflections with *I* > 2σ(*I*)

R_{int} = 0.030

θ_{max} = 74.5°, θ_{min} = 4.9°

h = -22→22

k = -8→8

l = -11→14

Refinement

Refinement on *F*²

Least-squares matrix: full

R [*F*² > 2σ(*F*²)] = 0.080

wR (*F*²) = 0.233

S = 1.23

2903 reflections

199 parameters

0 restraints

Primary atom site location: dual

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

w = 1/[σ²(*F_o*²) + (0.0572*P*)² + 6.0123*P*]

where *P* = (*F_o*² + 2*F_c*²)/3

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.47 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.38 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.6397 (2)	0.5834 (6)	0.8751 (3)	0.0220 (8)
C2	0.6287 (2)	0.5485 (6)	0.7528 (3)	0.0218 (8)
H2	0.669009	0.519350	0.711237	0.026*
C3	0.5568 (2)	0.5574 (5)	0.6928 (3)	0.0189 (7)
H3	0.547397	0.536452	0.608242	0.023*
C4	0.4979 (2)	0.5973 (5)	0.7558 (3)	0.0178 (7)
C5	0.5114 (2)	0.6345 (5)	0.8784 (3)	0.0201 (8)
H5	0.471482	0.664740	0.920495	0.024*
C6	0.5836 (2)	0.6276 (6)	0.9397 (3)	0.0234 (8)
H6	0.593767	0.652639	1.023700	0.028*
C7	0.42268 (19)	0.5993 (5)	0.6869 (3)	0.0188 (7)
C8	0.3054 (2)	0.5755 (5)	0.6523 (3)	0.0194 (8)
C9	0.22741 (19)	0.5532 (5)	0.6776 (3)	0.0188 (8)
C10	0.2034 (2)	0.7286 (6)	0.7450 (4)	0.0229 (8)
H10A	0.235907	0.742685	0.823383	0.027*
H10B	0.208148	0.843843	0.696746	0.027*
C11	0.1226 (2)	0.7041 (6)	0.7656 (4)	0.0249 (9)
H11	0.107002	0.816945	0.809285	0.030*
C12	0.0729 (2)	0.6855 (7)	0.6436 (4)	0.0280 (9)
H12A	0.020522	0.672769	0.656054	0.034*
H12B	0.077248	0.800506	0.594894	0.034*
C13	0.0958 (2)	0.5124 (6)	0.5766 (3)	0.0261 (9)
H13	0.063156	0.501154	0.496999	0.031*
C14	0.1769 (2)	0.5336 (6)	0.5561 (3)	0.0241 (8)
H14A	0.191844	0.421626	0.512660	0.029*
H14B	0.181982	0.646678	0.506007	0.029*
C15	0.1155 (2)	0.5267 (7)	0.8408 (4)	0.0282 (9)
H15A	0.147857	0.538200	0.919522	0.034*
H15B	0.063748	0.512872	0.855987	0.034*
C16	0.1378 (2)	0.3532 (6)	0.7743 (4)	0.0264 (9)
H16	0.132697	0.237763	0.823688	0.032*
C17	0.0881 (2)	0.3342 (6)	0.6516 (4)	0.0282 (9)
H17A	0.102861	0.221647	0.608557	0.034*
H17B	0.035805	0.318109	0.663834	0.034*
C18	0.2193 (2)	0.3748 (6)	0.7544 (4)	0.0242 (8)
H18A	0.234847	0.261751	0.712615	0.029*
H18B	0.251650	0.386012	0.833140	0.029*

F1	0.71024 (12)	0.5714 (4)	0.9363 (2)	0.0321 (6)
N1	0.40489 (17)	0.6139 (5)	0.5727 (3)	0.0237 (7)
N2	0.32721 (18)	0.5975 (5)	0.5493 (3)	0.0245 (7)
O1	0.36296 (13)	0.5751 (4)	0.7457 (2)	0.0192 (6)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0185 (18)	0.0220 (19)	0.0239 (19)	−0.0016 (15)	−0.0015 (14)	0.0036 (15)
C2	0.0179 (17)	0.0239 (19)	0.0245 (19)	−0.0014 (15)	0.0059 (14)	0.0014 (15)
C3	0.0218 (18)	0.0180 (17)	0.0177 (17)	0.0000 (14)	0.0055 (14)	−0.0012 (13)
C4	0.0152 (16)	0.0155 (16)	0.0232 (17)	−0.0024 (13)	0.0048 (13)	0.0008 (14)
C5	0.0199 (17)	0.0201 (18)	0.0217 (18)	0.0007 (14)	0.0083 (14)	0.0003 (14)
C6	0.028 (2)	0.0220 (19)	0.0204 (18)	−0.0023 (16)	0.0029 (15)	−0.0004 (15)
C7	0.0170 (17)	0.0187 (17)	0.0220 (18)	−0.0004 (14)	0.0074 (14)	0.0000 (14)
C8	0.0182 (17)	0.0217 (19)	0.0178 (17)	0.0005 (14)	0.0014 (13)	−0.0002 (14)
C9	0.0163 (17)	0.0260 (19)	0.0141 (16)	0.0021 (14)	0.0015 (13)	0.0001 (14)
C10	0.0163 (18)	0.027 (2)	0.0257 (19)	−0.0027 (15)	0.0034 (14)	−0.0046 (16)
C11	0.0153 (17)	0.034 (2)	0.027 (2)	0.0003 (16)	0.0061 (14)	−0.0057 (17)
C12	0.0159 (18)	0.039 (2)	0.029 (2)	0.0032 (17)	0.0029 (15)	0.0023 (18)
C13	0.0170 (18)	0.041 (2)	0.0191 (18)	−0.0018 (16)	−0.0021 (14)	−0.0039 (17)
C14	0.0179 (18)	0.038 (2)	0.0161 (17)	−0.0002 (16)	0.0018 (14)	−0.0001 (16)
C15	0.0197 (18)	0.045 (3)	0.0201 (19)	−0.0023 (17)	0.0047 (14)	−0.0018 (18)
C16	0.0197 (18)	0.031 (2)	0.030 (2)	−0.0026 (16)	0.0057 (15)	0.0047 (17)
C17	0.0205 (19)	0.034 (2)	0.030 (2)	−0.0060 (17)	0.0046 (16)	−0.0061 (18)
C18	0.0211 (18)	0.029 (2)	0.0233 (19)	0.0043 (16)	0.0051 (15)	0.0048 (16)
F1	0.0198 (11)	0.0465 (16)	0.0278 (12)	−0.0006 (10)	−0.0032 (9)	0.0031 (11)
N1	0.0194 (15)	0.0318 (18)	0.0207 (16)	0.0022 (13)	0.0061 (12)	0.0008 (14)
N2	0.0202 (16)	0.0366 (19)	0.0171 (15)	0.0020 (14)	0.0044 (12)	0.0013 (14)
O1	0.0148 (12)	0.0288 (14)	0.0146 (12)	−0.0007 (10)	0.0036 (9)	−0.0009 (10)

Geometric parameters (Å, °)

C1—C2	1.380 (5)	C11—H11	1.0000
C1—C6	1.376 (5)	C11—C12	1.532 (6)
C1—F1	1.369 (4)	C11—C15	1.529 (6)
C2—H2	0.9500	C12—H12A	0.9900
C2—C3	1.384 (5)	C12—H12B	0.9900
C3—H3	0.9500	C12—C13	1.528 (6)
C3—C4	1.400 (5)	C13—H13	1.0000
C4—C5	1.386 (5)	C13—C14	1.539 (5)
C4—C7	1.472 (5)	C13—C17	1.535 (6)
C5—H5	0.9500	C14—H14A	0.9900
C5—C6	1.394 (5)	C14—H14B	0.9900
C6—H6	0.9500	C15—H15A	0.9900
C7—N1	1.279 (5)	C15—H15B	0.9900
C7—O1	1.366 (4)	C15—C16	1.523 (6)
C8—C9	1.500 (5)	C16—H16	1.0000

C8—N2	1.288 (5)	C16—C17	1.538 (6)
C8—O1	1.369 (4)	C16—C18	1.544 (5)
C9—C10	1.550 (5)	C17—H17A	0.9900
C9—C14	1.534 (5)	C17—H17B	0.9900
C9—C18	1.548 (5)	C18—H18A	0.9900
C10—H10A	0.9900	C18—H18B	0.9900
C10—H10B	0.9900	N1—N2	1.408 (4)
C10—C11	1.537 (5)		
C6—C1—C2	123.8 (4)	H12A—C12—H12B	108.2
F1—C1—C2	118.3 (3)	C13—C12—C11	109.8 (3)
F1—C1—C6	117.9 (3)	C13—C12—H12A	109.7
C1—C2—H2	121.2	C13—C12—H12B	109.7
C1—C2—C3	117.5 (3)	C12—C13—H13	109.5
C3—C2—H2	121.2	C12—C13—C14	109.6 (3)
C2—C3—H3	119.8	C12—C13—C17	109.6 (3)
C2—C3—C4	120.5 (3)	C14—C13—H13	109.5
C4—C3—H3	119.8	C17—C13—H13	109.5
C3—C4—C7	117.6 (3)	C17—C13—C14	109.3 (3)
C5—C4—C3	120.2 (3)	C9—C14—C13	109.9 (3)
C5—C4—C7	122.2 (3)	C9—C14—H14A	109.7
C4—C5—H5	120.1	C9—C14—H14B	109.7
C4—C5—C6	119.9 (3)	C13—C14—H14A	109.7
C6—C5—H5	120.1	C13—C14—H14B	109.7
C1—C6—C5	118.1 (3)	H14A—C14—H14B	108.2
C1—C6—H6	121.0	C11—C15—H15A	109.7
C5—C6—H6	121.0	C11—C15—H15B	109.7
N1—C7—C4	127.2 (3)	H15A—C15—H15B	108.2
N1—C7—O1	113.1 (3)	C16—C15—C11	110.0 (3)
O1—C7—C4	119.7 (3)	C16—C15—H15A	109.7
N2—C8—C9	127.7 (3)	C16—C15—H15B	109.7
N2—C8—O1	112.5 (3)	C15—C16—H16	109.4
O1—C8—C9	119.8 (3)	C15—C16—C17	110.1 (3)
C8—C9—C10	110.7 (3)	C15—C16—C18	108.9 (3)
C8—C9—C14	107.7 (3)	C17—C16—H16	109.4
C8—C9—C18	111.2 (3)	C17—C16—C18	109.5 (3)
C14—C9—C10	109.3 (3)	C18—C16—H16	109.4
C14—C9—C18	109.1 (3)	C13—C17—C16	109.2 (3)
C18—C9—C10	108.9 (3)	C13—C17—H17A	109.8
C9—C10—H10A	109.8	C13—C17—H17B	109.8
C9—C10—H10B	109.8	C16—C17—H17A	109.8
H10A—C10—H10B	108.3	C16—C17—H17B	109.8
C11—C10—C9	109.3 (3)	H17A—C17—H17B	108.3
C11—C10—H10A	109.8	C9—C18—H18A	109.8
C11—C10—H10B	109.8	C9—C18—H18B	109.8
C10—C11—H11	109.4	C16—C18—C9	109.6 (3)
C12—C11—C10	109.2 (3)	C16—C18—H18A	109.8
C12—C11—H11	109.4	C16—C18—H18B	109.8

C15—C11—C10	109.7 (3)	H18A—C18—H18B	108.2
C15—C11—H11	109.4	C7—N1—N2	106.2 (3)
C15—C11—C12	109.6 (3)	C8—N2—N1	106.3 (3)
C11—C12—H12A	109.7	C7—O1—C8	102.0 (3)
C11—C12—H12B	109.7		
C1—C2—C3—C4	-1.1 (6)	C11—C15—C16—C17	-59.2 (4)
C2—C1—C6—C5	1.1 (6)	C11—C15—C16—C18	60.8 (4)
C2—C3—C4—C5	2.2 (6)	C12—C11—C15—C16	59.0 (4)
C2—C3—C4—C7	-178.3 (3)	C12—C13—C14—C9	59.4 (4)
C3—C4—C5—C6	-1.6 (6)	C12—C13—C17—C16	-59.5 (4)
C3—C4—C7—N1	-18.2 (6)	C14—C9—C10—C11	59.8 (4)
C3—C4—C7—O1	158.6 (3)	C14—C9—C18—C16	-59.3 (4)
C4—C5—C6—C1	0.0 (6)	C14—C13—C17—C16	60.6 (4)
C4—C7—N1—N2	176.5 (4)	C15—C11—C12—C13	-59.5 (4)
C4—C7—O1—C8	-176.9 (3)	C15—C16—C17—C13	59.3 (4)
C5—C4—C7—N1	161.3 (4)	C15—C16—C18—C9	-60.5 (4)
C5—C4—C7—O1	-22.0 (5)	C17—C13—C14—C9	-60.7 (4)
C6—C1—C2—C3	-0.5 (6)	C17—C16—C18—C9	59.9 (4)
C7—C4—C5—C6	178.9 (4)	C18—C9—C10—C11	-59.3 (4)
C7—N1—N2—C8	0.4 (4)	C18—C9—C14—C13	59.8 (4)
C8—C9—C10—C11	178.1 (3)	C18—C16—C17—C13	-60.5 (4)
C8—C9—C14—C13	-179.4 (3)	F1—C1—C2—C3	178.7 (3)
C8—C9—C18—C16	-177.9 (3)	F1—C1—C6—C5	-178.2 (3)
C9—C8—N2—N1	179.3 (4)	N1—C7—O1—C8	0.3 (4)
C9—C8—O1—C7	-179.6 (3)	N2—C8—C9—C10	-111.7 (4)
C9—C10—C11—C12	-60.3 (4)	N2—C8—C9—C14	7.7 (6)
C9—C10—C11—C15	59.8 (4)	N2—C8—C9—C18	127.1 (4)
C10—C9—C14—C13	-59.2 (4)	N2—C8—O1—C7	-0.1 (4)
C10—C9—C18—C16	59.8 (4)	O1—C7—N1—N2	-0.4 (4)
C10—C11—C12—C13	60.7 (4)	O1—C8—C9—C10	67.8 (4)
C10—C11—C15—C16	-60.9 (4)	O1—C8—C9—C14	-172.9 (3)
C11—C12—C13—C14	-60.0 (4)	O1—C8—C9—C18	-53.4 (4)
C11—C12—C13—C17	60.0 (4)	O1—C8—N2—N1	-0.2 (4)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C1—C6 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 \cdots N1 ⁱ	0.95	2.56	3.383 (5)	146
C18—H18A \cdots F1 ⁱⁱ	0.99	2.47	3.415 (5)	159

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+1, y-1/2, -z+3/2$.

2-(Adamantan-1-yl)-5-(4-chlorophenyl)-1,3,4-oxadiazole (II)

Crystal data

C₁₈H₁₉ClN₂O $M_r = 314.80$ Monoclinic, $P2_1/n$ $a = 13.08241$ (19) Å $b = 6.49259$ (9) Å $c = 18.5129$ (3) Å $\beta = 105.5609$ (16)° $V = 1514.83$ (4) Å³ $Z = 4$ $F(000) = 664$ $D_x = 1.380$ Mg m⁻³Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 11758 reflections

 $\theta = 3.7$ – 79.0 ° $\mu = 2.25$ mm⁻¹ $T = 160$ K

Needle, colourless

 $0.33 \times 0.12 \times 0.08$ mm

Data collection

XtaLAB Synergy, Dualflex, Pilatus 200K
diffractometerRadiation source: micro-focus sealed X-ray
tube, PhotonJet (Cu) X-ray Source

Mirror monochromator

Detector resolution: 5.8140 pixels mm⁻¹ ω scansAbsorption correction: analytical
(CrysAlisPro; Rigaku OD, 2019) $T_{\min} = 0.642$, $T_{\max} = 0.870$

14318 measured reflections

3217 independent reflections

3052 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.023$ $\theta_{\max} = 78.9$ °, $\theta_{\min} = 3.7$ ° $h = -16$ → 16 $k = -7$ → 8 $l = -23$ → 23

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.061$ $wR(F^2) = 0.167$ $S = 1.08$

3217 reflections

199 parameters

0 restraints

Primary atom site location: dual

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.094P)^2 + 1.5575P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.81$ e Å⁻³ $\Delta\rho_{\min} = -0.26$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.42626 (17)	0.6044 (4)	0.35491 (12)	0.0302 (5)
C2	0.40263 (17)	0.4056 (4)	0.37203 (12)	0.0317 (5)
H2	0.333320	0.351478	0.352048	0.038*
C3	0.48137 (17)	0.2855 (4)	0.41883 (12)	0.0279 (5)
H3	0.466239	0.148920	0.431206	0.034*
C4	0.58338 (16)	0.3685 (3)	0.44753 (11)	0.0254 (4)
C5	0.60437 (17)	0.5705 (3)	0.43076 (12)	0.0280 (5)
H5	0.672884	0.627166	0.451644	0.034*
C6	0.52611 (17)	0.6897 (4)	0.38380 (13)	0.0302 (5)

H6	0.540617	0.826921	0.371688	0.036*
C7	0.66623 (17)	0.2413 (3)	0.49555 (12)	0.0229 (4)
C8	0.81754 (16)	0.1694 (3)	0.56872 (11)	0.0238 (4)
C9	0.92972 (16)	0.2106 (3)	0.61176 (11)	0.0219 (4)
C10	0.93691 (16)	0.3930 (4)	0.66595 (12)	0.0282 (5)
H10A	0.897237	0.360738	0.703168	0.034*
H10B	0.904931	0.517124	0.637722	0.034*
C11	1.05425 (18)	0.4345 (4)	0.70636 (13)	0.0314 (5)
H11	1.059032	0.553003	0.741610	0.038*
C12	1.1016 (2)	0.2426 (4)	0.75071 (13)	0.0333 (5)
H12A	1.062078	0.209782	0.787976	0.040*
H12B	1.176566	0.268774	0.778004	0.040*
C13	1.09521 (18)	0.0597 (4)	0.69691 (13)	0.0319 (5)
H13	1.126743	-0.065180	0.726092	0.038*
C14	0.97843 (17)	0.0175 (3)	0.65594 (12)	0.0285 (5)
H14A	0.938650	-0.018439	0.692756	0.034*
H14B	0.973710	-0.100189	0.621181	0.034*
C15	1.11625 (17)	0.4853 (3)	0.64960 (14)	0.0319 (5)
H15A	1.191412	0.512639	0.676075	0.038*
H15B	1.086479	0.610423	0.621095	0.038*
C16	1.10922 (16)	0.3031 (4)	0.59555 (13)	0.0284 (5)
H16	1.149192	0.336599	0.557972	0.034*
C17	1.15702 (17)	0.1113 (4)	0.63970 (14)	0.0330 (5)
H17A	1.232475	0.136605	0.665944	0.040*
H17B	1.153464	-0.006047	0.604985	0.040*
C18	0.99249 (17)	0.2645 (3)	0.55507 (12)	0.0231 (4)
H18A	0.987062	0.149794	0.518988	0.028*
H18B	0.961848	0.389182	0.526636	0.028*
Cl1	0.32838 (4)	0.75434 (10)	0.29606 (3)	0.0384 (2)
N1	0.66468 (15)	0.0461 (3)	0.50869 (12)	0.0348 (5)
N2	0.76544 (15)	-0.0012 (3)	0.55744 (12)	0.0330 (4)
O1	0.76044 (11)	0.3299 (2)	0.53130 (8)	0.0254 (3)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0236 (10)	0.0423 (13)	0.0245 (10)	0.0028 (9)	0.0060 (8)	0.0015 (9)
C2	0.0215 (9)	0.0424 (13)	0.0291 (10)	-0.0040 (9)	0.0036 (8)	-0.0017 (9)
C3	0.0218 (10)	0.0354 (11)	0.0263 (10)	-0.0056 (8)	0.0060 (8)	-0.0019 (8)
C4	0.0218 (9)	0.0291 (11)	0.0248 (9)	-0.0027 (8)	0.0054 (7)	-0.0014 (8)
C5	0.0232 (9)	0.0290 (11)	0.0307 (10)	-0.0028 (8)	0.0050 (8)	-0.0010 (8)
C6	0.0256 (10)	0.0341 (11)	0.0310 (11)	-0.0006 (9)	0.0077 (8)	0.0012 (9)
C7	0.0186 (9)	0.0250 (10)	0.0244 (10)	-0.0048 (7)	0.0045 (8)	-0.0021 (7)
C8	0.0236 (9)	0.0217 (10)	0.0263 (9)	-0.0011 (8)	0.0069 (8)	-0.0003 (8)
C9	0.0200 (9)	0.0223 (9)	0.0233 (9)	-0.0016 (7)	0.0055 (7)	-0.0007 (7)
C10	0.0248 (10)	0.0286 (11)	0.0303 (10)	0.0010 (8)	0.0056 (8)	-0.0073 (8)
C11	0.0292 (11)	0.0290 (11)	0.0311 (11)	-0.0009 (9)	-0.0001 (8)	-0.0078 (9)
C12	0.0294 (11)	0.0431 (14)	0.0233 (10)	-0.0022 (9)	-0.0001 (9)	0.0021 (8)

C13	0.0289 (11)	0.0258 (11)	0.0349 (11)	0.0003 (9)	-0.0018 (9)	0.0083 (9)
C14	0.0285 (10)	0.0229 (10)	0.0314 (10)	-0.0029 (8)	0.0034 (8)	0.0052 (8)
C15	0.0238 (10)	0.0234 (10)	0.0432 (12)	-0.0052 (8)	-0.0004 (9)	0.0036 (9)
C16	0.0205 (10)	0.0332 (11)	0.0322 (11)	-0.0003 (8)	0.0081 (8)	0.0056 (9)
C17	0.0251 (10)	0.0296 (11)	0.0413 (12)	0.0057 (9)	0.0036 (9)	0.0000 (9)
C18	0.0213 (9)	0.0254 (10)	0.0225 (9)	-0.0001 (7)	0.0059 (8)	0.0018 (7)
C11	0.0242 (3)	0.0543 (4)	0.0342 (3)	0.0096 (2)	0.0034 (2)	0.0102 (2)
N1	0.0268 (9)	0.0279 (10)	0.0437 (11)	-0.0052 (7)	-0.0007 (8)	0.0024 (8)
N2	0.0258 (9)	0.0255 (9)	0.0426 (11)	-0.0055 (7)	0.0000 (8)	0.0036 (8)
O1	0.0203 (7)	0.0236 (7)	0.0300 (7)	-0.0035 (6)	0.0030 (6)	0.0005 (6)

Geometric parameters (Å, °)

C1—C2	1.384 (4)	C11—H11	1.0000
C1—C6	1.387 (3)	C11—C12	1.529 (3)
C1—C11	1.740 (2)	C11—C15	1.526 (3)
C2—H2	0.9500	C12—H12A	0.9900
C2—C3	1.393 (3)	C12—H12B	0.9900
C3—H3	0.9500	C12—C13	1.538 (3)
C3—C4	1.405 (3)	C13—H13	1.0000
C4—C5	1.391 (3)	C13—C14	1.537 (3)
C4—C7	1.460 (3)	C13—C17	1.532 (3)
C5—H5	0.9500	C14—H14A	0.9900
C5—C6	1.388 (3)	C14—H14B	0.9900
C6—H6	0.9500	C15—H15A	0.9900
C7—N1	1.292 (3)	C15—H15B	0.9900
C7—O1	1.360 (2)	C15—C16	1.536 (3)
C8—C9	1.494 (3)	C16—H16	1.0000
C8—N2	1.288 (3)	C16—C17	1.528 (3)
C8—O1	1.359 (3)	C16—C18	1.531 (3)
C9—C10	1.538 (3)	C17—H17A	0.9900
C9—C14	1.539 (3)	C17—H17B	0.9900
C9—C18	1.536 (3)	C18—H18A	0.9900
C10—H10A	0.9900	C18—H18B	0.9900
C10—H10B	0.9900	N1—N2	1.417 (3)
C10—C11	1.540 (3)		
C2—C1—C6	122.0 (2)	C11—C12—C13	109.73 (18)
C2—C1—C11	119.56 (17)	H12A—C12—H12B	108.2
C6—C1—C11	118.48 (19)	C13—C12—H12A	109.7
C1—C2—H2	120.3	C13—C12—H12B	109.7
C1—C2—C3	119.4 (2)	C12—C13—H13	109.5
C3—C2—H2	120.3	C14—C13—C12	109.36 (19)
C2—C3—H3	120.3	C14—C13—H13	109.5
C2—C3—C4	119.3 (2)	C17—C13—C12	109.38 (19)
C4—C3—H3	120.3	C17—C13—H13	109.5
C3—C4—C7	119.2 (2)	C17—C13—C14	109.69 (18)
C5—C4—C3	120.1 (2)	C9—C14—H14A	109.8

C5—C4—C7	120.68 (19)	C9—C14—H14B	109.8
C4—C5—H5	119.7	C13—C14—C9	109.39 (17)
C6—C5—C4	120.6 (2)	C13—C14—H14A	109.8
C6—C5—H5	119.7	C13—C14—H14B	109.8
C1—C6—C5	118.6 (2)	H14A—C14—H14B	108.2
C1—C6—H6	120.7	C11—C15—H15A	109.8
C5—C6—H6	120.7	C11—C15—H15B	109.8
N1—C7—C4	128.6 (2)	C11—C15—C16	109.40 (18)
N1—C7—O1	112.39 (19)	H15A—C15—H15B	108.2
O1—C7—C4	119.03 (17)	C16—C15—H15A	109.8
N2—C8—C9	129.94 (19)	C16—C15—H15B	109.8
N2—C8—O1	112.47 (18)	C15—C16—H16	109.5
O1—C8—C9	117.48 (17)	C17—C16—C15	109.55 (18)
C8—C9—C10	111.42 (17)	C17—C16—H16	109.5
C8—C9—C14	110.16 (17)	C17—C16—C18	109.88 (18)
C8—C9—C18	107.76 (17)	C18—C16—C15	108.81 (17)
C10—C9—C14	109.67 (17)	C18—C16—H16	109.5
C18—C9—C10	108.65 (17)	C13—C17—H17A	109.8
C18—C9—C14	109.12 (17)	C13—C17—H17B	109.8
C9—C10—H10A	109.8	C16—C17—C13	109.39 (18)
C9—C10—H10B	109.8	C16—C17—H17A	109.8
C9—C10—C11	109.27 (17)	C16—C17—H17B	109.8
H10A—C10—H10B	108.3	H17A—C17—H17B	108.2
C11—C10—H10A	109.8	C9—C18—H18A	109.6
C11—C10—H10B	109.8	C9—C18—H18B	109.6
C10—C11—H11	109.3	C16—C18—C9	110.41 (17)
C12—C11—C10	109.02 (19)	C16—C18—H18A	109.6
C12—C11—H11	109.3	C16—C18—H18B	109.6
C15—C11—C10	110.36 (18)	H18A—C18—H18B	108.1
C15—C11—H11	109.3	C7—N1—N2	105.90 (18)
C15—C11—C12	109.41 (19)	C8—N2—N1	106.14 (18)
C11—C12—H12A	109.7	C8—O1—C7	103.10 (16)
C11—C12—H12B	109.7		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C1—C6 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C3—H3 \cdots N1 ⁱ	0.95	2.61	3.386 (3)	139
C12—H12 <i>A</i> \cdots Cg1 ⁱⁱ	0.99	2.73	3.680 (3)	160

Symmetry codes: (i) $-x+1, -y, -z+1$; (ii) $x+1/2, -y+1/2, z+1/2$.